in that it is disubstituted at the atom $\alpha$ to the carbonyl C atom.

We acknowledge the support of the US National Science Foundation for the purchase of the Nicolet $R 3 m$ diffractometer and computing system.

## References

Clardy, J., Springer, J. P., Buchi, G., Matsuo, K. \& Wightman, R. (1975). J. Am. Chem. Soc. 97, 663-665.

Im, M. (1990). Private communication.
Lenstra, A. T. H., Petit, G. H. \& Geise, H. J. (1979). Cryst. Struct. Commun. 8, 1023-1029.
Sheldrick, G. M. (1983). SHELXTL Users Manual. Revision 5. Nicolet XRD Corporation, Madison, Wisconsin, USA.
Sinclair, P. J., Zhai, D., Reibenspies, J. \& Williams, R. M. (1986). J. Am. Chem. Soc. 108, 1103-1104.

Smith, J. L., Chwang, A. \& Sundaralingam, M. (1980). Acta Cryst. B36, 833-837.
Zhukhlistova, N. E., Smirnova, V. I., Tishchenko, G. N. \& Andrianov, V. I. (1977). Kristallografiya, 22, 77-81.
Zhukhlistova, N. E. \& Tischenko, G. N. (1980). Kristallografiya, 25, 274-279.

Acta Cryst. (1991). C47, 1998-2000

# Tricyclo[8.2.1.0 ${ }^{2,9}$ ]trideca-5,11-dien-13-one 

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(Received 19 November 1990; accepted 4 March 1991)


#### Abstract

C}_{13} \mathrm{H}_{16} \mathrm{O}, \quad M_{r}=188 \cdot 3\), orthorhombic, $P 2_{1} 2_{1} 2_{1}, \quad a=8.6255(12), \quad b=8.8372$ (12), $\quad c=$ 13.6741 (7) $\AA, \quad V=1042.3$ (3) $\AA^{3}, \quad Z=4, \quad D_{x}=$ $1.200 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda(\mathrm{Cu} K \alpha)=1.54184 \AA, \quad \mu=$ $5.36 \mathrm{~cm}^{-1}, F(000)=408, T=297 \mathrm{~K}, R=0.031$ for 1785 observations with $I>3 \sigma(I)$ (of 2138 unique data). The cyclooctene ring adopts the twist-boat conformation. There is a near-zero torsion angle of the boat at the ring-fusion bond, with magnitude of $-2.8(2)^{\circ}$. The torsion angles about the bonds comprising the sides of the boat are $33.8(2)$ and 18.5 (2) ${ }^{\circ}$. The cyclooctene and norbornenone $\mathrm{C}=\mathrm{C}$ bonds have lengths of 1.297 (3) and $1 \cdot 318$ (2) $\AA$, respectively, and the $\mathrm{C}=\mathrm{O}$ bond length is $1 \cdot 205$ (2) A.

Experimental. The title compound was prepared by allowing one equivalent of 1,1 -dimethoxy-2,3,4,5tetrachlorocyclopentadiene to react with eight equivalents of 1,5 -cyclooctadiene neat at $406-413 \mathrm{~K}$ (Akhtar, Fray \& Yarrow, 1968) followed by reduction in sodium-ethanol and hydrolysis with sulfuric acid-water-ether (Eaton, Sidhu, Langford, Cullison \& Pietruszewski, 1987).




[^0]Crystals that deposited from the reaction flask, m.p. 338-340 K, were suitable; a clear colorless crystal with dimensions $0.15 \times 0.18 \times 0.32 \mathrm{~mm}$ was used for data collection on an Enraf-Nonius CAD-4 diffractometer with $\mathrm{Cu} K \alpha$ radiation and a graphite monochromator. Cell dimensions were determined from setting angles of 25 reflections having $30>\theta>$ $25^{\circ}$. The $\omega-2 \theta$ scans were designed for $I=20 \sigma(I)$, subject to max. scan time $=60 \mathrm{~s}$, scan rates varied from $1 \cdot 0-4 \cdot 1^{\circ} \mathrm{min}^{-1}$. A hemisphere of data having 2 $<\theta<75^{\circ}, 0 \leq h \leq 10,-11 \leq k \leq 11,-17 \leq l \leq 17$ was measured and corrected for background, Lorentz, polarization and decay. $\psi$ scans of four reflections exhibited no decrease in intensity with rotation about the diffraction vector, thus no absorption correction was applied. Three standard reflections $(400,031,006)$ decreased in intensity by $5.0 \%$ apparently due to sublimation and a linear correction was applied. 4637 data were measured, equivalent data were averaged, $R_{\text {int }}=0.014$, yielding 2138 unique data. Systematic absences $h 00$ with $h$ odd, 0 k 0 with $k$ odd and $00 l$ with $l$ odd indicated space group $P 2_{1} 2_{1} 2_{1}$. The structure was solved by direct methods using RANTAN (Yao, 1981), refined by full-matrix least squares based upon $F$, using data for which $I>$ $3 \sigma(I)$, weights $w=4 F_{o}^{2}\left[\sigma^{2}(I)+\left(0.02 F_{o}^{2}\right)^{2}\right]^{-1}$ using the Enraf-Nonius Structure Determination Package (Frenz \& Okaya, 1980), scattering factors of Cromer \& Waber (1974), and anomalous coefficients of Cromer (1974). Heavy-atom coordinates were refined with anisotropic thermal parameters; H -atom coordinates were located by $\Delta F$ and were refined with
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Table 1. Coordinates and equivalent isotropic thermal parameters

| $B_{\mathrm{cq}}=\left(8 \pi^{2} / 3\right) \sum_{i} \sum_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$ |  |  |  |  |
| :--- | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $B_{\text {eq }}\left(\AA^{2}\right)$ |
|  | $x$ | $0.4059(2)$ | $0.89767(9)$ | $6.72(3)$ |
| O | $0.6674(1)$ | $0.5348(1)$ | $0.81568(9)$ | $3.42(2)$ |
| C1 | $0.3318(1)$ | $0.5000(2)$ | $0.92635(9)$ | $3.61(2)$ |
| C2 | $0.2987(1)$ | $0.4894(2)$ | $0.9581(1)$ | $4.83(3)$ |
| C3 | $0.1300(2)$ | $0.6448(2)$ | $0.9835(2)$ | $7.57(5)$ |
| C4 | $0.0595(2)$ | $0.7797(2)$ | $0.9280(1)$ | $5.75(4)$ |
| C5 | $0.1116(2)$ | $0.7908(2)$ | $0.8336(2)$ | $5.94(4)$ |
| C6 | $0.1216(2)$ | $0.6693(2)$ | $0.7647(1)$ | $6.81(4)$ |
| C7 | $0.0758(2)$ | $0.5496(2)$ | $0.7437(1)$ | $5.03(3)$ |
| C8 | $0.1974(2)$ | $0.4033(2)$ | $0.7847(1)$ | $3.99(3)$ |
| C9 | $0.4429(2)$ | $0.2571(2)$ | $0.7908(1)$ | $4.69(3)$ |
| C10 | $0.3550(2)$ | $0.2279(2)$ | $0.8835(1)$ | $5.26(3)$ |
| C11 | $0.3259(2)$ | $0.3518(2)$ | $0.9446(1)$ | $4.62(3)$ |
| C12 | $0.3929(2)$ | $0.3885(2)$ | $0.8797(1)$ | $4.42(3)$ |
| C13 | $0.5319(2)$ |  |  |  |

Table 2. Bond distances $(\AA)$, angles $\left({ }^{\circ}\right)$ and selected torsion angles ( ${ }^{\circ}$ )

| O-C13 | $1 \cdot 205$ (2) | C5-C6 | $1 \cdot 297$ (3) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C} 1-\mathrm{C} 2$ | 1.571 (2) | C6-C7 | 1.483 (3) |
| $\mathrm{C} 1-\mathrm{C} 8$ | 1.526 (2) | C7-C8 | 1.517 (2) |
| $\mathrm{Cl}-\mathrm{C} 9$ | 1.565 (2) | C9-C10 | 1.500 (2) |
| C2-C3 | 1.522 (2) | C9-C13 | 1.514 (2) |
| $\mathrm{C} 2-\mathrm{Cl} 2$ | 1.561 (2) | $\mathrm{Cl} 0-\mathrm{Cl} 1$ | 1.318 (2) |
| C3-C4 | 1.542 (2) | $\mathrm{Cl1-Cl2}$ | 1.494 (2) |
| C4-C5 | 1.483 (3) | C12-Cl3 | 1.526 (2) |
| $\mathrm{C} 2-\mathrm{Cl}-\mathrm{C} 8$ | $120 \cdot 0$ (1) | $\mathrm{Cl}-\mathrm{C} 9-\mathrm{Cl0}$ | 108.4 (1) |
| $\mathrm{C} 2-\mathrm{Cl}-\mathrm{C} 9$ | $103 \cdot 1$ (1) | $\mathrm{Cl}-\mathrm{C} 9-\mathrm{Cl} 3$ | 98.2 (1) |
| C8-Cl-C9 | $110 \cdot 8$ (1) | C10-C9-C13 | 97.8 (1) |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | 117.5 (1) | C9- $\mathrm{C} 10-\mathrm{Cl1}$ | 108.5 (1) |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{Cl2}$ | 102.9 (1) | $\mathrm{C} 10-\mathrm{Cl1-Cl2}$ | 108.7 (1) |
| C3-C2-C12 | 113.6 (1) | $\mathrm{C} 2-\mathrm{C} 12-\mathrm{Cl1}$ | 108.9 (1) |
| $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | 112.7 (1) | $\mathrm{C} 2-\mathrm{C} 12-\mathrm{Cl} 3$ | 97.9 (1) |
| C3-C4-C5 | 118.8 (2) | $\mathrm{Cl1}-\mathrm{Cl2}-\mathrm{Cl} 3$ | 97.7 (1) |
| C4-C5-C6 | 126.2 (2) | O-Cl3-C9 | 131.0 (1) |
| C5-C6-C7 | 124.0 (2) | $\mathrm{O}-\mathrm{Cl} 3-\mathrm{Cl2}$ | $132 \cdot 1$ (2) |
| C6-C7-C8 | $116 \cdot 1$ (1) | $\mathrm{C} 9-\mathrm{Cl} 3-\mathrm{Cl} 2$ | $96 \cdot 8$ (1) |
| $\mathrm{Cl}-\mathrm{C} 8-\mathrm{C} 7$ | 117.6 (1) |  |  |
| C8-C1-C2-C3 | -2.8(2) | $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | -85.5 (2) |
| C2-C1-C8-C7 | 58.4 (2) | $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5$ | $33 \cdot 8$ (2) |
| C6-C7-C8-Cl | 18.5 (2) | C4-C5-C6--C7 | 2.5 (3) |
| C3-C4-C5-C6 | 48.7 (3) | C5-C6-C7-C8 | -85.0 (2) |

isotropic thermal parameters. Final $R=0.031$ for 1785 observed data ( 0.041 for all 2138 data), $w R=$ 0.037 , and $S=1.456$ for 192 variables. Max. shift $0.09 \sigma$ in the final cycle, max. residual density 0.18 , $\min .-0.13 \mathrm{e} \AA^{-3}$, and extinction coefficient $g=$ $9.34(12) \times 10^{-6}$ where the factor $\left(1+g I_{c}\right)^{-1}$ was applied to $F_{c}$. The fractional coordinates of the title compound are given in Table 1. Fig. 1 is a perspective drawing showing the atom numbering. Fig. 2 illustrates the unit cell. Bond distances, angles, and selected torsion angles are presented in Table 2.*

[^1]Related literature. Bond lengths and bond angles $\mathrm{C} 1-\mathrm{C} 2 \quad 1.571(2), \quad \mathrm{C} 10-\mathrm{Cl1} \quad 1.318$ (2) $\AA$ and C8-Cl—C9 $110 \cdot 8(1)^{\circ}$ of tricyclo[8.2.1.0 ${ }^{2,9}$ ]trideca-5,11-dien-13-one are similar to those of endo,-endo,anti-1,6,7,8,9,14,15,16-octachloropentacyclo[12.2.1.1 $\left.{ }^{6,9} .0^{2,13} .0^{5,10}\right]$ octadeca- 7,15 -diene $\quad[\mathrm{C} 1-\mathrm{C} 2$ 1.587 (3), $\mathrm{C} 10-\mathrm{C} 111.327$ (3) $\AA$ and $\mathrm{C} 8-\mathrm{Cl}-\mathrm{C} 9$ $114 \cdot 1$ (2) ${ }^{\circ}$ ] (Garcia, Fronczek \& McLaughlin, 1991a) and to those of endo,endo,anti-1,6,7,8,9,14,15,-16-octachloro-17,17,18,18-tetramethoxypentacyclo[12.2.1.1 ${ }^{6,9} .0^{2,13} .0^{5,19}$ ]octadeca- 7,15 -diene $\quad[\mathrm{C} 1-\mathrm{C} 2$ 1.579 (3), $\mathrm{C} 10-\mathrm{C} 11 \quad 1.324(3) \AA$ and $\mathrm{C} 8-\mathrm{Cl}-\mathrm{C} 9$ 113.2 (2) ${ }^{\circ}$ ] (Garcia, Fronczek \& McLaughlin, 1991b). The C5-C6 bond distance of $1-297$ (3) $\AA$ for the title compound is in agreement with that of trans-spiro[bicyclo[6.3.0]undec-4-ene-10,2'-[1,3]-dioxolane]-2,2,7,7-tetracarboxylate, $\quad 1 \cdot 319$ (4) $\AA$ (Bertram, Jansen, Peters, Meier \& Winterfeldt, 1986). Torsion angle $\mathrm{C} 8-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ at the ring fusion bond is $-2.8(2)^{\circ}$ in accordance with those of endo,-endo,anti-1,6,7,8,9,14,15,16-octachloropentacyclo[12.2.1.1 ${ }^{6,9} .0^{2,13} .0^{5,10}$ ]octadeca-7,15-diene [-0.3(3) ${ }^{\circ}$ ] (Garcia et al., 1991a), and endo,endo,anti-1,6,7,8,9,14,15,16-octachloro-17,17,18,18-tetramethoxypentacyclo $\left[12.2 .1 .1^{6,9} \cdot 0^{2,13} .0^{5,10}\right]$ octadeca- $7,15-$ diene $\left[1.8(3)^{\circ}\right]$ (Garcia et al., 1991b).


Fig. 1. ORTEP (Johnson, 1965) drawing of the molecule, representing $C$ atoms as $30 \%$ probability ellipsoids and $H$ atoms as circles of arbitrary radius.


Fig. 2. Stereoview of the unit cell. $\mathbf{c}$ is vertical, $\mathbf{b}$ is horizontal and $\mathbf{a}$ is into the plane of the paper.

The cyclooctene ring has a twist-boat conformation, with torsion angles $\omega_{1} \mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 1-\mathrm{C} 8, \mathrm{C} 4-$ C5-C6-C7, $\omega_{2} \mathrm{C} 2-\mathrm{C} 1-\mathrm{C} 8-\mathrm{C} 7, \mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3-$ $\mathrm{C} 4, \mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6, \mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ and $\omega_{3}$ $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5, \mathrm{C} 1-\mathrm{C} 8-\mathrm{C} 7-\mathrm{C} 6$ of $\omega_{1}=$ $-2.8(2), \omega_{2}=58.4$ (2), -85.5 (2) and $\omega_{3}=33.8$ (2), $18.5(2)^{\circ}$ that are distorted from the ideal boat $D_{2 d}$ conformation (of cylooctane) having torsion angles of $\omega_{1}= \pm 0, \quad \omega_{2}= \pm 76.4$ and $\omega_{3}= \pm 0^{\circ}$ (Hendrickson, 1964), and $\omega_{1}= \pm 0, \omega_{2}= \pm 73 \cdot 4$, and $\omega_{3}= \pm 0^{\circ}$ (Ivanov \& Osawa, 1984).

We thank the Louisiana Educational Quality Support Fund Grant No. LEQSF(1987-90)-RD-A-5 for support of this research.

## References

Akhtar, I. A., Fray, G. I. \& Yarrow, J. M. (1968). J. Chem. Soc. C, pp. 812-815.

Bertram, H.-J., Jansen, M., Peters, K., Meier, A. \& Winterfeldt, E. (1986). Justus Liebigs Ann. Chem. pp. 456-464.
Cromer, D. T. (1974). International Tables for X-ray Crystallography, Vol. IV, Table 2.3.1. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht.)
Cromer, D. T. \& Waber, J. T. (1974). International Tables for $X$-ray Crystallography, Vol. IV, Table 2.2B. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht.)
Eaton, P. A., Sidhu, R. S., Langford, G. E., Cullison, D. A. \& Pietruszewski, C. L. (1987). Tetrahedron, 37, 4479-4493.
Frenz, B. A. \& Okaya, Y. (1980). Enraf-Nonius Structure Determination Package. Enraf-Nonius, Delft, The Netherlands.
Garcia, J. G., Fronczek, F. R. \& McLaughlin, M. L. (1991a). Acta Cryst. C47, 451-453.
Garcia, J. G., Fronczek, F. R. \& McLaughlin, M. L. (1991b). Acta Cryst. C47, 206-209.
Hendrickson, J. B. (1964). J. Am. Chem. Soc. 86, 48544866.

Ivanov, P. M. \& Osawa, E. (1984). J. Comput. Chem. 5(4), 307-313.
Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
Yao, J.-X. (1981). Acta Cryst. A37, 642-644.

Acta Cryst. (1991). C47, 2000-2002

# 5-Methyl-4-(1-piperidyl)-5-trifluoromethyl-1,2-oxathiole 

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(Received 12 November 1990; accepted 18 February 1991)

Abstract. $\quad \mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~F}_{3} \mathrm{NO}_{3} \mathrm{~S}, \quad M_{r}=285 \cdot 28$, orthorhombic, $P c a 2_{1}, a=9.324$ (2), $b=12.038$ (3),$c=$ $11.252(3) \AA, \quad V=1263.0(5) \AA^{3}, \quad Z=4, \quad D_{x}=$ $1.50 \mathrm{~g} \mathrm{~cm}^{-3}$, Мо $K \alpha, \lambda=0.71069 \AA, \mu=2.97 \mathrm{~cm}^{-1}$, $F(000)=592, \quad T=291 \mathrm{~K}, \quad R=0.032 \quad$ for 1124 observed reflections. The five-membered unsaturated sultone ring is slightly nonplanar adopting a flat C2 half-chair conformation with endocyclic torsion angles of $5.2(8),-1.9(8),-1.8(8), 5.1(8)$ and $-6.6(8)^{\circ}$. The endocyclic $\mathrm{S}-\mathrm{O}$ and $\mathrm{S}-\mathrm{C}$ bond lengths are 1.599 (2) and 1.689 (4) $\AA$, respectively, and the $\mathrm{C}-\mathrm{S}-\mathrm{O}$ angle is $95 \cdot 4$ (2) ${ }^{\circ}$.

Experimental. The reaction of methanesulfonyl chloride with the sodium alcoholate of the 2-hydroxythioamide (1), in the presence of an excess of sodium hydride, leads unexpectedly to the product
(2) (Maliverney, Merenyi \& Viehe, 1990). This compound was identified as being a sultone, i.e. a cyclic sulfonate and the X-ray analysis was undertaken in order to confirm this structure and to determine the geometry of this new unsaturated heterocycle.


$$
\begin{aligned}
& \text { A: } \mathrm{NaH} 2 \mathrm{eq} ., \mathrm{THF}, 30 \mathrm{~min}, 293 \mathrm{~K} \\
& \mathrm{~B}: \mathrm{Cl}-\mathrm{SO}_{2}-\mathrm{CH}_{3} 1 \cdot 2 \text { eq., } 24 \mathrm{~h}, 293 \mathrm{~K}
\end{aligned}
$$


(2)

Parallelepiped crystal of approximate dimensions $0.35 \times 0.22 \times 0.14 \mathrm{~mm}$. Lattice parameters refined
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[^0]:    * Author to whom correspondence should be addressed

[^1]:    * Lists of structure factors, anisotropic thermal parameters, distances and angles involving H atoms, torsion angles, and H -atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54058 ( 17 pp. ). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

